

Addition/Correction

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A Valence Bond Modeling of Trends in Hydrogen Abstraction Barriers and Transition States of Hydroxylation Reactions Catalyzed by Cytochrome P450 Enzymes

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A Valence Bond Modeling of Trends in Hydrogen Abstraction Barriers and Transition States of Hydroxylation Reactions Catalyzed by Cytochrome P450 Enzymes [*J. Am. Chem. Soc.* 2008, *130*, 10128– 10140]. Sason Shaik,* Devesh Kumar, and Sam P. de Visser*

Page 10134. Figure 4 was plotted with the wrong labels on the axes. The corrected figure is given below. We thank Prof. James Mayer for pointing this error out to us.

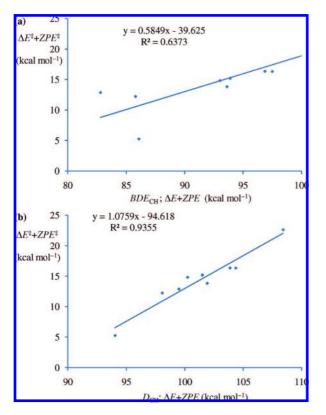


Figure 4. (a) A plot of the average barriers $\Delta E^{\ddagger} + ZPE^{\ddagger}$ (B2) for the quartet and doublet H-abstraction steps vs BDE_{CH} (B2). (b) A plot of the average barriers $\Delta E^{\ddagger} + ZPE^{\ddagger}$ for the quartet and doublet H-abstraction steps vs D_{CH} (defined by eq 3) as calculated with basis set B2.

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10.1021/ja806917f Published on Web 09/20/2008 An Expedient Asymmetric Synthesis of Platencin [*J. Am. Chem. Soc.* 2008, *130,* 11292–11293]. K. C. Nicolaou,* Qiao-Yan Toh, and David Y.-K. Chen*

Page 11293. Reference 15 should read as follows:

(15) Alcohol **11**, derived from the major diastereoisomer of the Diels-Alder reaction, oxidized faster than the minor diastereoisomer, facilitating the differentiation of these isomers. A single recrystallization (hexane) of the resulting ketone gave essentially enantiomerically pure material (by optical rotation measurement).

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